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ON THE SYNTHESIS OF 4-KETO-STEROIDAL ALKALOIDS

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ABSTRACT.—To obtain 4-keto-steroidal alkaloids from solasodine, two routes were tried: allylic acetoxylation of (22*S*,25*R*)-22,26-*N*-Cbz-epiminocholest-5-ene-3 β ,16 β -diol-acetate [3]; and hydroboration of (22*S*,25*R*)-16 β -acetyl-22,26-*N*-Cbz-epiminocholest-4-en-3-one [11]. The first route yielded (22*S*,25*R*)-3 β -hydroxy-16 β -acetoxy-22,26-*N*-Cbz-epiminocholestan-5,6-oxido-4-one [10]. The second one yielded two products: (22*S*,25*R*)-3 β -hydroxy-16 β -ethoxy-22,26-*N*-Cbz-epimino-5 α -cholestan-4-one [22] and its 16 β -acetoxy homologue [23].

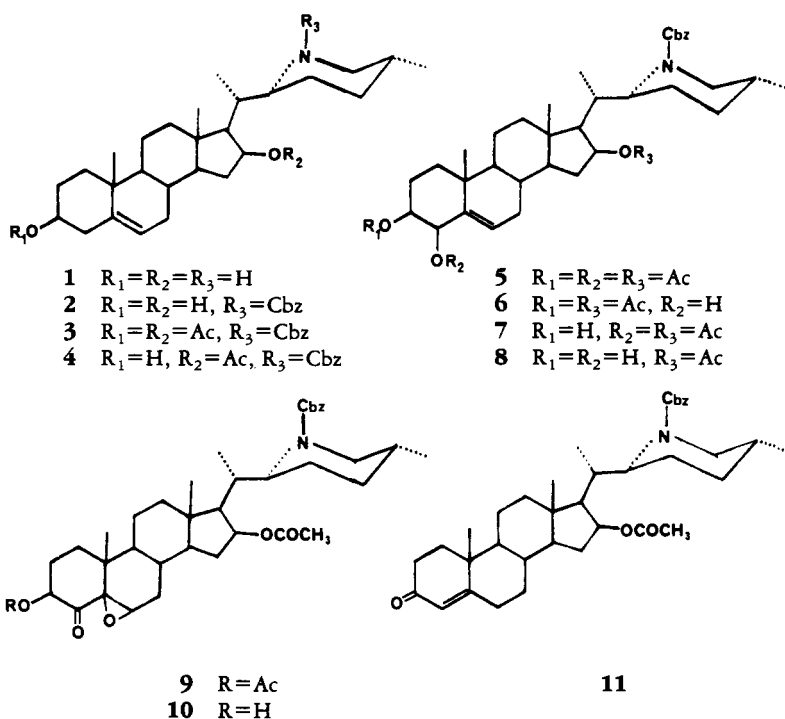
Steroidal alkaloids with a 3-hydroxy-4-keto moiety on ring A have been found in two species of Solanaceae, *Solanum oblongifolium* Bitter (1) and *Solanum ecuadorensis* Bitter (2). The interesting pharmacological properties of these alkaloids (3) led us to consider the convenience of obtaining them by synthesis. As a first approach to the problem, the introduction of a 4-keto group in the steroid nucleus was tried. Solasodine was used as starting material because it was available and is easily transformed into an epiminocholestane type alkaloid with a lateral chain where the nitrogen atom is α -oriented, as in solaphyllidine (4) and related alkaloids (5).

Allylic oxidation of cholesteryl acetate using SeO₂ has been reported by Rosenheim and Starling (6) with rather low yields. On the other hand Pb(OAc)₄ oxidation of cholest-5-en-3-one (7) introduces a 4-equatorial hydroxyl group which is difficult to oxidize. Therefore, it was considered convenient to try allylic acetoxylation (8) to introduce a 4-axial hydroxyl.

Reduction of solasodine with NaBH₄ (9) produced (22*S*,25*R*)-22,26-epiminocholest-5-ene-3 β ,16 β -diol [1]. To protect the amino group, 1 was treated with benzyl chloroformate (Cbz). It is interesting to note that the aromatic ring of the Cbz moiety interacts with the H-26eq, deshielding it strongly, causing this proton to appear at δ 3.80 in the ¹H-nmr spectrum. The H-26eq is coupled to the H-26ax ($J = 14$ Hz), and the latter is also coupled to the H-25ax ($J = 5$ Hz). Finally, the H-25ax is coupled to the C-27 methyl. Decoupling experiments established these relationships, and a similar situation is present in certain sapogenins (10).

To obtain a compound amenable to allylic acetoxylation, the *N*-Cbz derivative [2] was treated with Ac₂O/C₅H₅N to yield the 3,16-diacetoxy derivative [3]. Treatment with Br₂ and silver acetate in CHCl₃ at low temperature produced a mixture of starting material 3 and three products: (22*S*,25*R*)-*N*-Cbz-22,26-epiminocholest-5-ene-3 β ,4 β ,16 β -triol-acetate [5], (22*S*,25*R*)-3 β ,16 β -di-*O*-acetyl-22,26-*N*-Cbz-epiminocholest-5-en-4 β -ol [6], and its 4 β ,16 β -di-*O*-acetyl-3 β -ol isomer 7. Unfortunately the desired product 6 was obtained in lowest yield; therefore 5, 7, and a mixture of 6 and 7 were subjected to mild hydrolysis with K₂CO₃ in MeOH, conditions that do not affect the acetate on C-16. The 3 β ,4 β -diol 8 was obtained almost quantitatively. Selective acetylation of 8 with Ac₂O/C₅H₅N at 0° gave 6 in 54% yield. Mild oxidation of 6 with Jones reagent yielded 9, which, upon treatment with K₂CO₃/MeOH at room temperature, gave 10.

Even though 10 is an interesting compound, the desired product was an alkaloid with a 3 β -OH, 4-keto moiety with no neighboring epoxy group. Therefore, a new scheme was tried to obtain a compound devoid of the Δ^5 double bond. To obtain such a compound, 3 was treated with K₂CO₃/MeOH, and the product 4 of hydrolysis was subjected to Oppenauer oxidation to yield 11, which is a suitable compound to intro-



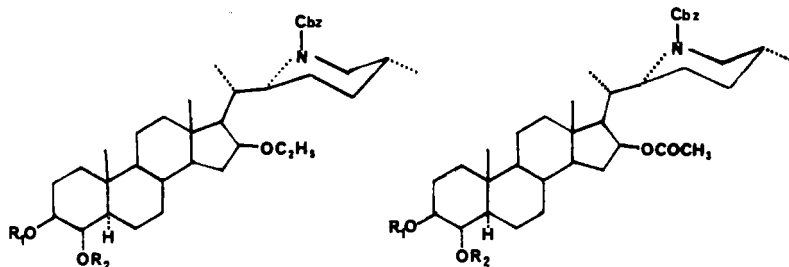
duce a hydroxyl at C-4 by hydroboration. The hydroboration reaction, which was performed according to Zweifel and Herbert (11), rendered two products in about equal proportions. The less polar product **12**, R_f 0.27, showed an ethoxy group at C-16, while the other product **16**, R_f 0.16, had an *O*-acetyl. The formation of **12** was probably caused by an excess of diborane, as reduction of carbonyls under these conditions has been reported in the literature (12).

In order to oxidize the hydroxyl at C-4 without affecting the one at C-3, selective acetylation of **12** and **16** was conducted as previously described for **8**. In each case three acetylated derivatives were obtained: **13**, **14** and **15** from **12**, and **17**, **18**, and **19** from **16**. Mild oxidation of **14** with Jones reagent yielded the 3-*O*-acetyl-4-keto derivative **20**. The ^{13}C -nmr spectrum of **20** showed the signal of C-3 at 76.2 ppm, that of C-5 at 57.4 ppm, and the carbonyl (C-4) at 205.3 ppm, which agrees with values reported for solaphyllidine acetate (13). In a similar manner, **21** was obtained by mild oxidation of **18**.

For comparison, **15** was treated with Jones reagent, yielding **24**. In this compound C-4 appears at 77.2 ppm and C-5 at 51.5 ppm. Treatment of **20** and **21** with $K_2CO_3/MeOH$ produced the corresponding 3-OH, 4-keto derivatives **22** and **23**. These compounds show the C-4 carbonyl at 212.4 ppm as observed in solaphyllidine.

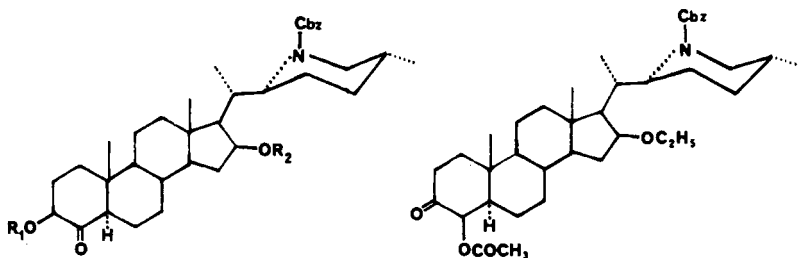
Because catalytic reduction could not be used to cleave the Cbz group (14), because it would reduce the carbonyl, acid hydrolysis with $HBr/HOAc$ (15) was tried unsuccessfully. On the other hand, removal of the Cbz group using a strong Lewis acid such as $AlCl_3$ (16) proved equally inadequate. This demonstrated the convenience of using a protecting group like *t*-butyloxycarbonyl (BOC) (17), which can be removed by acid hydrolysis (18), or thiothiazolone (19).

The overall yield for compound **10** was approximately 7%, while the yield of **23** from solasodine was about 4%. The yield of the latter could be doubled if partial reduction of the 16-*O*-acetate during hydroboration is avoided; this could be accomplished if this group is hydrolyzed prior to hydroboration.



- 12 $R_1=R_2=H$
 13 $R_1=R_2=Ac$
 14 $R_1=Ac, R_2=H$
 15 $R_1=H, R_2=Ac$

- 16 $R_1=R_2=H$
 17 $R_1=R_2=Ac$
 18 $R_1=Ac, R_2=H$
 19 $R_1=H, R_2=Ac$



- 20 $R_1=Ac, R_2=Et$
 21 $R_1=R_2=Ac$
 22 $R_1=H, R_2=Et$
 23 $R_1=H, R_2=Ac$

24

EXPERIMENTAL

GENERAL EXPERIMENTAL PROCEDURES.—Tlc was performed on Si gel plates, using C_6H_{14} -EtOAc (2:1) as a solvent unless otherwise stated, and spots were visualized with I_2 vapor. Melting points were determined on a Fisher-Johns hot stage and are uncorrected. Optical rotations were measured in a Schmidt-Haensch polarimeter. The ir spectra were recorded using a Perkin-Elmer spectrometer model Fx-1720 as KBr disks. The 1H -nmr and ^{13}C -nmr spectra were determined in $CDCl_3$ with TMS as internal standard, and chemical shifts are expressed in ppm; a Varian Ft-80 apparatus was used, and standard proton noise-decoupled and attached proton test (APT) spectra were recorded for all compounds. Microanalyses were conducted at Prof. Melissa & G. Reuter's Analytical Laboratories, D-5270 Gummersbach, Germany. Solasodine dihydrochloride was obtained from Oss-Diosynth (Holland).

(2*S*,25*R*)-22,26-EPIMINOCHOLEST-5-ENE-3 β ,16 β -DIOL [1].—To an ice-cold solution of 60 g (0.145 mol) of solasodine in 3.5 liters of MeOH/ CH_2Cl_2 , 22 g (0.58 mol) of $NaBH_4$ was added slowly with stirring. After 2 h, ice- H_2O was added and the mixture was extracted twice with $CHCl_3$. The organic phase was dried over anhydrous Na_2SO_4 and concentrated under vacuum to dryness. The crude product was crystallized from MeOH to give 53.1 g (88%) of dihydrosolasodine: mp 259–262°, [lit. (20) 265–266°]; $ir \nu_{max} cm^{-1}$ 3410, 3190, 1170, 1080; 1H nmr (80 MHz, $CDCl_3$) δ 0.85 (3H, d, $J=6$ Hz, 27-Me), 0.94 (3H, s, Me-18), 1.03 (3H, s, Me-19), 1.08 (3H, d, $J=6$ Hz, Me-21), 2.95 (1H, d, $J=12$ Hz, H-26), 3.45 (1H, m, H-3 α), 4.40 (1H, m, H-16 α), 5.30 (1H, bd, $J=4$ Hz, H-6).

(2*S*,25*R*)-*N*-CBZ-22,26-EPIMINOCHOLEST-5-ENE-3 β ,16 β -DIOL [2].—A solution of 45 g (0.108 mol) of **1** in 1.3 liters of $CHCl_3$ was mixed with 647 ml of 5% $NaHCO_3$ and 34 g (0.22 mol) of benzyl chloroformate (Cbz-Cl) in toluene, prepared according to Carter *et al.* (21). After shaking for 9.5 h an additional amount (11 g) of CbzCl was added and left overnight at room temperature. The $CHCl_3$ phase was shaken several times with H_2O , dried over anhydrous Na_2SO_4 , and concentrated to dryness. The residue was chromatographed on a Si gel column. Elution with C_6H_6 -EtOAc (10:1) yielded 34.75 g (58.4%) of **2**, which crystallized from EtOH- Me_2CO (1:1): mp 170–171°; $[\alpha]^{25}_D = -14.5$ ($c=0.31$, MeOH). Calcd for $C_{35}H_{51}NO_4$, C 76.46, H 9.35, N 2.55; found C 76.09, H 9.11, N 2.44. $Ir \nu_{max}$ 3440, 1693 cm^{-1} ; 1H

nmr (80 MHz, CDCl₃) δ 0.84 (3H, d, $J = 6$ Hz, Me-21), 0.86 (3H, s, Me-18), 0.98 (3H, s, Me-19), 1.00 (3H, d, $J = 6$ Hz, Me-27), 3.01 (1H, dd, $J = 14, 5$ Hz, H-26ax), 3.80 (1H, d, $J = 14$ Hz, H-26eq), 4.12 (1H, m, H-16 α), 5.10 (2H, ABq, $J = 22, 12$ Hz, H-Bz), 7.34 (5H, bs, aromatic H).

(22*S*, 25*R*)-*N*-Cbz-22,26-EPIMINOCHOLEST-5-ENE-3 β , 16 β -DIOL-ACETATE (**3**).—To a solution of 30 g of **2** in pyridine, 150 ml of Ac₂O was added. After 3 days at room temperature cold H₂O was added, and the precipitate was filtered, washed with H₂O, and crystallized from MeOH, yielding 34 g of diacetate: mp 110–112°; ir ν max 1732, 1686, 1245 cm⁻¹; ¹H nmr (80 MHz, CDCl₃) δ 0.84 (3H, s, Me-18), 0.90 (3H, d, $J = 6$ Hz, Me-21), 0.96 (3H, d, $J = 6$ Hz, Me-27), 1.04 (3H, s, Me-19), 2.00 (3H, s, OAc), 2.04 (3H, s, OAc), 4.6 (1H, m, H-3 α), 5.07 (2H, ABq, $J = 23, 13$ Hz, H₂-Bz), 5.26 (1H, bs, H-4), 7.35 (5H, bs, aromatic H).

ALLYLIC ACETOXYLATION OF **3**.—A solution of **3** (10 g, 0.016 mol) in 71 ml CHCl₃ was cooled at -60° and mixed under stirring with 0.64 g (0.008 mol) of Br₂ and a solution of 14.25 g (0.085 mol) AgAc in pyridine. The mixture was stirred until it reached ambient temperature and then left to stand 24 h in the dark. The mixture was treated with dilute HCl to complete precipitation of AgCl, filtered, and washed with H₂O. The aqueous layer was extracted with CHCl₃, and the CHCl₃ phase was shaken with NaHCO₃ solution, dried over anhydrous Na₂SO₄, and concentrated to dryness. The residue (9 g) was purified by vacuum chromatography using Si gel. Elution with C₆H₁₄-EtOAc (10:1) yielded 1.59 g of starting compound **3** and 0.88 g of **5**, which crystallized out of the column solvent: mp 85–88°; [α]²⁵_D -31.6° ($c = 0.0155$, MeOH) ir ν max cm⁻¹ 1745, 1690, 1245; ¹H nmr (80 MHz, CDCl₃) δ 0.84 (3H, s, Me-18), 0.90 (3H, d, $J = 6$ Hz, Me-21), 0.98 (3H, d, $J = 6$ Hz, Me-27), 1.18 (3H, s, Me-19), 1.98 (6H, s, OAc), 2.03 (3H, s, OAc), 3.06 (1H, dd, $J = 14, 5$ Hz, H-26ax), 3.80 (1H, d, $J = 14$ Hz, H-26eq), 4.75 (1H, m, H-3 α), 5.08 (2H, ABq, $J = 23, 13$ Hz), 5.50 (1H, d, $J = 4$ Hz, H-6), 5.75 (1H, m, H-4 α), 7.3 (5H, bs, aromatic H); ¹³C nmr (20 MHz, CDCl₃) ppm 138.5 (C-5), 131.0 (C-6), 76.9 (C-4), 76.5 (C-16), 72.8 (C-3), 56.4 (C-22), 55.8 (C-17), 54.4 (C-14), 50.0 (C-9), 45.9 (C-26), 42.5 (C-13), 39.5 (C-12), 36.8 (C-1), 36.1 (C-20), 36.0 (C-10), 34.8 (C-15), 31.6 (C-7), 31.1 (C-8), 28.8 (C-25), 27.4 (C-23), 25.4 (C-24), 22.5 (C-2), 20.2 (C-11 and C-27), 18.9 (C-19), 13.8 (C-21), 12.6 (C-18). Elution with C₆H₁₄-EtOAc (4:1) yielded 0.08 g of **6**: mp 87–90°; [α]²⁵_D -20.3 ($c = 0.04$ MeOH). Calcd for C₃₉H₅₅NO₇, C 72.07, H 8.53, N 2.16; found C 71.90, H 8.17, N 2.14. Ir ν max cm⁻¹ 3456, 1735, 1695, 1245; ¹H nmr (80 MHz, CDCl₃) δ 0.82 (3H, s, Me-18), 0.93 (3H, d, $J = 6$ Hz, Me-21), 0.98 (3H, d, $J = 6$ Hz, Me-27), 1.18 (3H, s, Me-19), 2.00 (6H, s, OAc), 2.09 (3H, s, OAc), 3.02 (1H, dd, $J = 14, 5$ Hz, H-26ax), 3.80 (1H, d, $J = 14$ Hz, H-26eq), 4.20 (1H, d, $J = 4$ Hz, H-4 α), 4.70 (1H, m, H-3 α), 5.10 (2H, ABq, $J = 22, 12$), 5.70 (1H, m, H-6), 7.3 (5H, bs, aromatic H); ¹³C nmr see Table 1. Increasing the polarity of the solvent gave a 4.5 g mixture of **6** and **7**. Finally, 0.78 g of **7**, mp 82–85°, was obtained: [α]²⁵_D -31.4 ($c = 0.27$, MeOH); ir ν max cm⁻¹ 3450, 1735, 1694, 1244; ¹H nmr (80 MHz, CDCl₃) δ 0.83 (3H, s, Me-18), 0.94 (3H, d, $J = 6$ Hz, Me-21), 0.98 (3H, d, $J = 6$ Hz, Me-27), 1.10 (3H, s, Me-19), 3.02 (1H, dd, $J = 14, 5$ Hz, H-26ax), 3.40 (1H, m, H-3 α), 3.80 (1H, d, $J = 14$ Hz, H-26eq), 5.10 (2H, ABq, $J = 2, 12$, H-Bz), 5.35 (1H, m, H-4 α), 5.70 (1H, d, $J = 4$ Hz, H-6), 7.30 (5H, bs, aromatic H); ¹³C nmr see Table 1.

HYDROLYSIS OF ALLYLIC ACETOXYLATION PRODUCTS.—To a mixture of **5**, **7**, and **6** + **7** dissolved in MeOH, a 5% solution of K₂CO₃ was added to slight turbidity and the mixture left at ambient temperature for 6 h. After addition of H₂O the product was extracted with CHCl₃. The organic phase was dried over anhydrous Na₂SO₄ and concentrated to dryness. Crystallization from MeOH afforded 4.7 g of **8**: mp 168–172°; [α]²⁵_D -11.2 ($c = 0.0187$, MeOH). Calcd for C₃₇H₅₃NO₆, C 73.11, H 8.79, N 2.30; found C 73.15, H 8.73, N 2.31. Ir ν max cm⁻¹ 3474, 1735, 1695, 1243; ¹H nmr (80 MHz, CDCl₃) δ 0.84 (3H, s, Me-18), 0.90 (3H, d, $J = 6$ Hz, Me-21), 0.95 (3H, d, $J = 6$ Hz, Me-27), 1.12 (3H, s, Me-19), 1.97 (3H, s, OAc), 2.99 (1H, dd, $J = 14, 5$ Hz, H-26ax), 3.40 (1H, m, H-3 α), 3.76 (1H, d, $J = 14$ Hz, H-26eq), 4.05 (1H, d, $J = 4$ Hz, H-4 α), 5.10 (2H, ABq, $J = 23, 13$ Hz, H-Bz), 5.65 (1H, m, H-6), 7.28 (5H, bs, aromatic H); ¹³C nmr (20 MHz, CDCl₃) ppm 143.0 (C-5), 127.9 (C-6), 76.6 (C-4 and C-17), 72.4 (C-3), 56.4 (C-22), 55.9 (C-17), 54.5 (C-14), 50.2 (C-9), 45.9 (C-26), 42.6 (C-13), 39.5 (C-12), 37.0 (C-1), 36.0 (C-10), 34.8 (C-15), 31.6 (C-7 and C-8), 28.7 (C-25), 27.4 (C-23), 26.0 (C-2), 25.3 (C-24), 20.8 (C-11 and C-19), 19.9 (C-27), 13.8 (C-21), 12.6 (C-18).

SELECTIVE ACETYLTATION OF **8**.—To an ice-cold solution of 3.7 g of **8** in 25 ml of pyridine, 20 ml of Ac₂O was added. After 35 min cold H₂O was added, and the mixture was left at ambient temperature for 1 h. The precipitate was filtered, washed with H₂O, and dried overnight in an oven at 45°. The residue was dissolved in toluene and taken to dryness under vacuum, and the products were separated by vacuum chromatography on Si gel. Elution with C₆H₁₄-EtOAc (3:1) yielded 0.79 g of **5**. Further elution with C₆H₁₄-EtOAc (2:1) yielded 2.08 g of **6**.

(22,25*R*)-3 β , 16 β -DIACETOXY-22,26-*N*-Cbz-EPIMINOCHOLESTAN-5,6-OXIDO-4-ONE (**9**).—To

TABLE 1. ¹³C Chemical Shifts of N-Cbz-dihydrosolasodine [2] and Derivatives.*

Carbon	Compound																						
	2	3	6	7	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24				
C-1	37.3	38.1	37.0	36.9	38.0	34.8	37.9	37.3	36.1	37.7	37.9	37.2	36.0	37.4	36.0	36.0	35.9	35.8	38.8				
C-2	31.6	27.8	21.7	31.8	30.3	33.8	31.5	26.0	25.8	31.3	31.6	25.6	25.7	31.1	28.5	28.5	32.2	32.4	31.0				
C-3	71.6	73.8	75.4	71.6	65.2	198.6	75.5	74.0	79.3	74.4	75.4	74.0	79.3	74.3	76.2	76.1	74.5	74.4	204.7				
C-4	42.3	37.0	75.3	79.1	204.7	123.0	76.7	75.4	72.5	78.4	76.7	75.4	72.5	78.3	205.3	205.0	212.2	212.1	77.2				
C-5	141.0	139.5	141.8	139.0	69.5	170.4	50.9	49.4	51.5	49.2	50.9	49.1	51.5	49.2	57.4	57.4	56.9	56.8	51.5				
C-6	121.1	122.2	128.4	130.0	63.2	32.6	22.9	22.9	27.7	22.9	23.0	22.4	22.6	23.0	21.5	21.4	21.5	21.3	24.1				
C-7	31.6	31.4	31.6	31.5	29.8	31.7	31.2	31.2	31.4	29.1	31.0	30.8	31.2	29.3	30.1	29.9	30.2	30.1	29.1				
C-8	31.5	31.3	31.2	31.1	34.2	35.1	34.5	34.7	34.8	34.7	34.5	34.5	34.5	34.5	34.5	34.7	34.0	33.9	34.7				
C-9	50.1	50.1	50.2	50.2	48.6	53.6	53.9	58.9	53.9	54.0	53.0	54.9	51.5	54.0	53.7	54.2	54.0	53.8	53.8				
C-10	36.5	36.6	36.1	36.0	46.2	37.0	36.5	36.1	36.8	36.2	36.3	36.0	36.8	36.0	42.6	42.8	43.2	43.1	36.8				
C-11	20.7	20.7	20.2	20.2	21.2	20.7	20.8	20.8	20.8	20.8	21.2	21.2	20.7	20.7	20.2	20.1	20.2	20.1	21.0				
C-12	39.9	39.6	39.5	39.5	39.3	39.5	39.8	39.9	39.9	39.9	39.9	39.2	39.7	39.7	39.9	39.6	40.0	39.8	39.9				
C-13	42.3	42.6	42.6	42.5	42.9	42.6	42.7	42.2	42.2	42.3	42.8	42.8	42.8	42.7	42.3	42.3	42.4	42.4	43.0				
C-14	54.4	54.4	54.5	54.5	54.1	53.6	54.3	54.2	54.4	54.3	54.3	54.1	54.2	54.3	54.3	53.8	54.3	54.0	54.0				
C-15	37.3	34.8	34.8	34.7	34.7	34.8	33.6	33.6	33.6	33.6	33.8	34.8	34.7	34.8	33.6	34.4	33.7	34.4	33.6				
C-16	73.0	76.6	76.5	76.4	75.9	76.3	79.6	79.6	79.6	79.6	76.4	76.5	76.6	76.6	76.6	76.4	79.6	76.3	79.6				
C-17	57.0	55.9	55.9	55.8	55.9	55.8	58.3	58.2	58.2	58.2	55.9	55.9	55.9	55.9	58.2	55.9	58.2	55.9	58.2				
C-18	13.2	12.6	12.6	12.6	12.6	12.7	12.8	13.0	12.8	13.0	12.8	12.8	12.8	12.8	13.0	12.8	13.0	12.8	12.9				
C-19	19.3	19.1	20.9	20.2	18.2	17.3	13.5	13.3	13.5	13.4	13.5	13.3	13.5	13.4	13.6	13.4	13.4	13.4	13.7				
C-20	35.6	36.0	35.9	35.9	36.4	35.7	36.2	35.9	36.1	36.2	35.9	36.0	36.0	36.0	36.1	36.0	36.0	36.0	36.2				
C-21	14.2	13.8	13.8	13.8	13.7	13.7	13.8	13.5	13.8	13.7	13.8	13.8	13.8	13.8	13.8	13.8	13.8	13.8	13.7				
C-22	56.6	56.5	56.4	56.4	56.5	56.5	56.5	56.3	56.4	56.3	56.5	56.5	56.4	56.5	56.3	56.5	56.3	56.5	56.2				
C-23	27.7	27.4	27.4	27.4	27.6	27.4	27.8	27.9	27.8	27.9	27.8	27.4	27.4	27.4	27.6	27.4	27.6	27.8	27.8				
C-24	24.9	25.4	25.4	25.3	25.7	25.4	25.5	25.8	25.8	25.8	25.7	25.4	25.5	25.5	25.8	25.5	25.8	25.6	25.8				
C-25	28.9	28.8	28.7	28.7	29.0	28.7	28.9	29.0	29.1	29.4	28.7	28.8	28.8	28.8	29.1	28.8	29.0	28.9	29.1				
C-26	46.1	45.9	45.9	45.8	46.1	45.9	46.0	46.1	46.0	46.1	45.9	45.9	45.9	45.9	46.1	45.9	46.0	46.1	46.1				
C-27	19.9	19.9	19.9	19.9	20.3	19.9	20.0	20.4	20.4	20.4	19.9	20.0	20.0	20.3	20.4	20.1	20.4	20.2	20.4				

*Other signals: 67.1 (Bz-CH₂), 127.9, 128.1, 128.5, 137.0 (Aromatic-C), 157.6 (Bz-CO), Acetate 169.9–170.0 (CO), 20.6–21.1 (Me), Etoxy group 63.8 (CH₂-O), 15.6 (Me).

a cold solution of 300 mg of **6** in 50 ml of Me₂CO, 1.0 ml of Jones reagent was added (drop by drop). After 4 h at room temperature, MeOH and H₂O were added, the product was extracted with CHCl₃, and the organic phase was washed with H₂O, dried with anhydrous Na₂SO₄, and evaporated to dryness. The residue showed two spots on tlc which were separated by preparative tlc on Si gel using three developments with C₆H₁₄-EtOAc (2:1). The upper layer yielded 240 mg of **9** as a yellow powder, mp 95–97°. Calcd for C₃₉H₅₃NO₈, C 70.56, H 8.05, N 2.11; found C 70.39, H 8.27, N 2.25. Ir ν max cm⁻¹ 1734, 1695, 1242; ¹H nmr (80 MHz, CDCl₃) δ 0.85 (3H, s, Me-18), 0.88 (3H, d, *J* = 6 Hz, Me-21), 0.94 (3H, d, *J* = 6 Hz, Me-27), 0.98 (3H, s, Me-19), 2.00 (3H, s, OAc), 2.08 (3H, s, OAc), 3.00 (1H, dd, *J* = 14, 5 Hz, H-26ax), 3.30 (1H, d, *J* = 3 Hz, H-6), 3.78 (1H, d, *J* = 14 Hz, H-26eq), 4.35 (1H, m, H-3 α), 5.10 (2H, ABq, *J* = 23, 13, H-Bz), 5.29 (1H, m, H-16 α), 7.30 (5H, bs, aromatic H); ¹³C nmr (20 MHz, CDCl₃) ppm 204.0 (C-4), 76.2 (C-17), 68.8 (C-3), 67.7 (C-5), 60.6 (C-6), 56.4 (C-22), 55.8 (C-17), 54.1 (C-14), 49.9 (C-9), 46.2 (C-10), 46.0 (C-26), 42.6 (C-13), 39.4 (C-12), 37.0 (C-1), 34.5 (C-15), 33.8 (C-8), 32.5 (C-7), 28.8 (C-2 and C-25), 27.4 (C-23), 25.5 (C-24), 21.1 (C-11), 20.0 (C-27), 17.5 (C-19), 13.7 (C-21), 12.5 (C-18).

(22*S*,25*R*)-3 β -HYDROXY-16 β -ACETOXY-22,26-*N*-Cbz-EPIMINO-CHOLESTAN-5,6-OXIDO-4-ONE [**10**].—A solution of 80 mg of **9** in 50 ml MeOH was saturated with a 5% K₂CO₃ solution and left at room temperature 24 h. Extraction with CHCl₃ and usual workup gave 65 mg of **10** which did not crystallize: ir ν max cm⁻¹ 1732, 1272; ¹H nmr (80 MHz, CDCl₃) δ 0.83 (3H, s, Me-18), 0.87 (3H, d, *J* = 6 Hz, Me-21), 0.93 (3H, d, *J* = 6 Hz, Me-27), 1.05 (3H, s, Me-19), 2.99 (1H, dd, *J* = 14, 5, H-26ax), 3.25 (1H, d, *J* = 4 Hz, H-6), 3.75 (1H, d, *J* = 14 Hz, H-26eq), 3.90 (1H, m, H-3 α), 5.10 (2H, ABq, *J* = 22, 12, H-Bz), 5.25 (1H, m, H-16 α), 7.30 (5H, bs, aromatic H); ¹³C nmr see Table 1.

(22*S*,25*R*)-16 β -ACETOXY-22,26-*N*-Cbz-EPIMINOCHOLEST-5-EN-3 β -OL [**4**].—A solution of 6.3 g (0.01 mol) of **3** in 280 ml MeOH was saturated with 18 ml of a 5% K₂CO₃ solution and left at room temperature overnight. Extraction with CHCl₃ and usual workup yielded 5.2 g (0.0088 mol) of **4**, which crystallized from EtOH; mp 176–178; ir ν max cm⁻¹ 3447, 1734, 1699, 1241; ¹H nmr (80 MHz, CDCl₃) δ 0.82 (3H, s, Me-18), 0.91 (3H, d, *J* = 6 Hz, Me-21), 0.95 (3H, d, *J* = 6 Hz, Me-27), 0.99 (3H, s, Me-19), 1.99 (3H, s, OAc), 3.00 (1H, dd, *J* = 14, 5, H-26ax), 3.40 (1H, m, H-3 α), 3.80 (1H, dd, *J* = 14 Hz, H-26eq), 5.10 (2H, ABq, *J* = 22, 12 Hz, H-Bz), 5.32 (1H, m, H-6), 7.3 (5H, bs, aromatic H).

(22*S*,25*R*)-16 β -ACETOXY-22,26-*N*-Cbz-EPIMINOCHOLEST-4-EN-3-ONE [**11**].—A mixture of 3.0 g of **4**, 135 ml of toluene, and 27 ml of cyclohexanone was heated to complete solution. Aluminum isopropoxide (1.8 g) was added, and 100 ml of toluene was distilled. H₂O was added and the product extracted with CH₂Cl₂. Chromatography over alumina yielded 1.95 g of **11**: mp 112–115°; [α]²⁵_D 39.0 (*c* = 0.031, MeOH). Calcd for C₃₇H₅₁NO₅, C 75.35, H 8.72, N 2.37; found C 75.10, H 8.59, N 2.35. Ir ν max cm⁻¹ 1732, 1693; ¹H nmr (80 MHz, CDCl₃) δ 0.83 (3H, s, Me-18), 0.90 (3H, d, *J* = 6 Hz, Me-21), 0.95 (3H, d, *J* = 6 Hz, Me-27), 1.12 (3H, s, Me-19), 1.97 (3H, s, OAc), 2.95 (1H, dd, *J* = 14, 5 Hz, H-26ax), 3.75 (1H, d, *J* = 14 Hz, H-26eq), 5.07 (2H, ABq, *J* = 22, 12, H-Bz), 5.25 (1H, m, H-16 α), 5.64 (1H, bs, H-4), 7.30 (5H, bs, aromatic H); ¹³C nmr see Table 1.

HYDROBORATION OF **11**.—The BF₃-etherate complex and THF were distilled under N₂, and the diglyme was purified (22) just before use. A solution of 1.5 g of **11** in 20 ml dry THF was put in contact with a stream of N₂ containing B₂H₆ generated separately according to Zweifel and Herbert (11); 27 ml of 1.0 M NaBH₄ in diglyme was added drop by drop to a solution of BF₃ complex (6.9 ml) in diglyme (6 ml). After the addition of NaBH₄ was complete, the generator was heated to 70° to complete the transference of B₂H₆. The organo-borane complex was treated with 20 ml 3 N NaOH and 20 ml of 30% H₂O₂ under stirring. The mixture was extracted with Et₂O. The organic phase was saturated with NaCl and extracted twice with 25 ml of Et₂O. The Et₂O layer was dried over anhydrous MgSO₄ and concentrated under vacuum. The residue (1.3 g) showed two substances on tlc with R_f 0.27 and 0.16. Separation was accomplished on Si gel using vacuum chromatography. C₆H₆-EtOAc (5:1) eluted **12** (510 mg): mp 144–146°; [α]²⁵_D 16.4 (*c* = 0.011, MeOH). Calcd for C₃₇H₅₇NO₅, C 74.58, H 9.64, N 2.35; found C 74.43, H 9.62, N 2.36. Ir ν max cm⁻¹ 3519, 1696, 1262, 1080; ¹H nmr (80 MHz, CDCl₃) δ 0.78 (3H, s, Me-18), 0.83 (3H, s, Me-19), 1.03 (3H, d, *J* = 6 Hz, Me-21), 1.14 (3H, d, *J* = 6 Hz, Me-27), 3.02 (1H, dd, *J* = 14, 5 Hz, H-26ax), 3.40 (1H, m, H-3 α), 3.50 (1H, m, H-4 α), 3.85 (1H, d, *J* = 14 Hz, H-26eq), 3.95 (1H, m, H-16), 5.10 (2H, ABq, *J* = 22, 12, H-Bz), 7.3 (5H, bs, aromatic H); ¹³C nmr see Table 1. Increasing the polarity of the solvent to C₆H₆-EtOAc (2:1) yielded 435 mg of **16**: mp 103–104°; [α]²⁵_D 27.0 (*c* = 0.0133, MeOH). Calcd for C₃₇H₅₅NO₆·MeOH, C 71.10, H 9.27, N 2.18; found C 70.86, H 8.91, N 2.19. Ir ν max cm⁻¹ 3447, 1734, 1690, 1266, 1247, 1064; ¹H nmr (80 MHz, CDCl₃) δ 0.76 (3H, s, Me-18), 0.82 (3H, s, Me-19), 0.89 (3H, d, *J* = 6 Hz, Me-21), 0.98 (3H, d, *J* = 6 Hz, Me-27), 3.04 (1H, dd, *J* = 14, 5 Hz, H-26ax), 3.35 (1H, m, H-3 α), 3.50 (1H, m, H-4 α), 3.80 (1H, d, *J* = 14 Hz, H-26eq), 5.10 (2H, ABq, *J* = 22, 12 Hz, H-Bz), 5.25 (1H, m, H-16 α), 7.3 (5H, bs, aromatic H); ¹³C nmr see Table 1.

SELECTIVE ACETYLTATION OF 12.—A solution containing 400 mg of **12** in 3.0 ml of pyridine at 2° was reacted with Ac₂O for 15 min. Cold H₂O was added, and the precipitate was washed with H₂O, dried at 45°, dissolved in toluene, and taken to dryness under vacuum. A mixture (360 mg) of three products was obtained which was separated using vacuum chromatography over Si gel. C₆H₁₄/EtOAc (10:1) eluted **13** as a gum: [α]²⁵_D 20.7 (c = 0.015, MeOH). Calcd for C₄₁H₆₁NO₇, C 72.44, H 9.04, N 2.06; found C 72.17, H 9.29, N 2.18. Ir ν max cm⁻¹ 1742, 1697, 1250; ¹H nmr (80 MHz, CDCl₃) δ 0.76 (3H, s, Me-18), 0.84 (3H, s, Me-19), 0.98 (3H, d, J = 6 Hz, Me-21), 1.15 (3H, d, J = 6 Hz, Me-27), 1.98 (3H, s, OAc), 3.05 (1H, dd, J = 14, 5 Hz, H-26ax), 3.78 (1H, d, J = 14 Hz, H-26eq), 3.97 (1H, m, H-16α), 4.79 (1H, m, H-4α), 4.90 (1H, m, H-3α), 5.09 (2H, ABq, J = 22, 12, H-Bz), 7.3 (5H, bs, aromatic H); ¹³C nmr see Table 1. Increasing the polarity of the solvent yielded **14** (120 mg, R_f 0.66) as a gum. Calcd for C₃₉H₅₉NO₆, C 73.43, H 9.32, N 2.20; found C 73.65, H 9.20, N 2.07. Ir ν max cm⁻¹ 3452, 1734, 1697, 1250; ¹H nmr (80 MHz, CDCl₃) δ 0.75 (3H, s, Me-18), 0.84 (3H, s, Me-19), 1.02 (3H, d, J = 6 Hz, Me-21), 1.18 (3H, d, J = 6 Hz, Me-27), 2.00 (3H, s, OAc), 3.05 (1H, dd, J = 14, 5, H-26ax), 3.50 (1H, m, H-4α), 3.80 (1H, d, J = 14 Hz, H-26eq), 3.95 (1H, m, H-16α), 4.70 (1H, m, H-3α), 5.10 (2H, ABq, J = 23, 13, H-Bz), 7.3 (5H, bs, aromatic H); ¹³C nmr see Table 1. Finally, **15** was obtained, which could not be induced to crystallize. Calcd for C₃₉H₅₉NO₆, C 73.46, H 9.33, N 2.20; found C 73.55, H 9.47, N 2.15. Ir ν max cm⁻¹ 3450, 1734, 1695, 1248; ¹H nmr (80 MHz, CDCl₃) δ 0.75 (3H, s, Me-18), 0.83 (3H, s, Me-19), 1.01 (3H, d, J = 6 Hz, Me-21), 1.19 (3H, d, J = 6 Hz, Me-27), 2.05 (3H, s, OAc), 3.05 (1H, dd, J = 14, 5, H-26ax), 3.40 (1H, m, H-3α), 3.75 (1H, d, J = 14 Hz, H-26eq), 3.95 (1H, m, H-16α), 4.70 (1H, m, H-4α), 5.10 (2H, ABq, J = 23, 13 Hz, H-Bz), 7.30 (5H, bs, aromatic H); ¹³C nmr see Table 1.

SELECTIVE ACETYLTATION OF 16.—Acetylation was performed in a manner similar to the acetylation of **12**. The mixture of acetates was separated by vacuum chromatography on Si gel using C₆H₁₄/EtOAc. Compound **17** eluted first, [α]²⁵_D 13.6 (c = 0.006, MeOH). Calcd for C₄₁H₅₉NO₈, C 70.97, H 8.57, N 2.02; found C 70.85, H 8.40, N 1.90. Ir ν max cm⁻¹ 1734, 1695, 1250; ¹H nmr (80 MHz, CDCl₃) δ 0.76 (3H, s, Me-18), 0.86 (3H, s, Me-19), 0.89 (3H, d, J = 6 Hz, Me-21), 0.98 (3H, d, J = 6 Hz, Me-27), 1.95 (3H, s, OAc), 2.00 (3H, s, OAc), 3.00 (1H, dd, J = 14, 5 Hz, H-26ax), 3.76 (1H, d, J = 14 Hz, H-26eq), 4.70 (1H, m, H-3α), 4.85 (1H, m, H-4α), 5.10 (2H, ABq, J = 23, 13 Hz, H-Bz), 5.25 (1H, m, H-16), 7.3 (5H, bs, aromatic H); ¹³C nmr see Table 1. Increasing the polarity of the solvent eluted **18**. Calcd for C₃₉H₅₇NO₇, C 71.86, H 8.81, N 2.15; found C 71.64, H 8.72, N 2.04. Ir ν max cm⁻¹ 1735, 1694, 1245; ¹H nmr (80 MHz, CDCl₃) δ 0.76 (3H, s, Me-18), 0.80 (3H, s, Me-19), 0.90 (3H, d, J = 6 Hz, Me-21), 0.93 (3H, d, J = 6 Hz, Me-27), 1.95 (3H, s, O-Ac), 2.05 (3H, s, O-Ac), 3.00 (1H, dd, J = 14, 5 Hz, H-26ax), 3.42 (1H, m, H-4α), 3.78 (1H, d, J = 14 Hz, H-26eq), 4.6 (1H, m, H-3α), 5.10 (2H, ABq, J = 23, 13 Hz, H-Bz), 5.25 (1H, m, H-16α), 7.3 (5H, bs, aromatic H); ¹³C nmr see Table 1. Finally **19** was eluted: ir ν max cm⁻¹ 3447, 1737, 1695, 1242; ¹H nmr (80 MHz, CDCl₃) δ 0.75 (3H, s, Me-18), 0.85 (3H, s, Me-19), 0.90 (3H, d, J = 6 Hz, Me-21), 0.95 (3H, d, J = 6 Hz, Me-27), 1.99 (3H, s, OAc), 2.06 (3H, s, OAc), 2.98 (1H, dd, J = 14, 5 Hz, H-26ax), 3.37 (1H, m, H-3α), 3.75 (1H, d, J = 14 Hz, H-26eq), 4.80 (1H, m, H-4α), 5.10 (2H, ABq, J = 23, 13 Hz, H-Bz), 5.25 (1H, m, H-16α), 7.3 (5H, aromatic H); ¹³C nmr see Table 1.

(22S,25R)-3β-ACETOXY-16β-ETHOXY-22,26-N-Cbz-EPIMINO-5α-CHOLESTAN-4-ONE [**20**].—To a cold solution of **14** (80 mg) in Me₂CO, Jones reagent was added drop by drop to slight excess. The reaction mixture was left at room temperature for 1 h. After addition of MeOH and H₂O, the mixture was extracted with CHCl₃. The organic phase was dried over anhydrous Na₂SO₄ and taken to dryness under vacuum. Purification by vacuum chromatography over Si gel yielded 52 mg of **20**: mp 110–112°; [α]²⁵_D 0.6 (c = 0.017, MeOH). Calcd for C₃₉H₅₇NO₆, C 73.66, H 9.04, N 2.20; found C 73.49, H 9.08, N 2.35. Ir ν max cm⁻¹ 1749, 1733, 1696, 1236; ¹H nmr (80 MHz, CDCl₃) δ 0.72 (3H, s, Me-18), 0.78 (3H, s, Me-19), 0.88 (3H, d, J = 6 Hz, Me-21), 1.01 (3H, d, J = 6 Hz, Me-27), 2.05 (3H, s, OAc), 3.01 (1H, dd, J = 14, 5 Hz, H-26ax), 3.75 (1H, d, J = 14 Hz, H-26eq), 3.95 (1H, m, H-16α), 4.95 (1H, m, H-3α), 5.10 (2H, ABq, J = 23, 13 Hz, H-Bz), 7.3 (5H, bs, aromatic H); ¹³C nmr see Table 1.

(22S,25R)-4β-ACETOXY-16β-ETHOXY-22,26-N-Cbz-EPIMINO-5-CHOLESTAN-3-ONE [**24**].—Compound **15** (50 mg) was oxidized with Jones reagent in a similar manner to that described for **14** to yield 35 mg of **24**: ir ν max cm⁻¹ 1749, 1733, 1696, 1238; ¹H nmr (80 MHz, CDCl₃) δ 0.72 (3H, s, Me-18), 0.76 (3H, s, Me-19), 0.90 (3H, d, J = 7 Hz), 1.03 (3H, d, J = 7 Hz), 2.09 (3H, s, OAc), 3.04 (1H, dd, J = 14, 5 Hz, H-26ax), 3.76 (1H, d, J = 14 Hz, H-26eq), 3.95 (1H, m, H-16α), 5.05 (1H, m, H-4α), 5.10 (2H, ABq, J = 23, 13 Hz, H-Bz), 7.30 (5H, bs, aromatic H); ¹³C nmr see Table 1.

(22S,25R)-3β,16β-DIACETOXY-22,26-N-Cbz-EPIMINO-5α-CHOLESTAN-4-ONE [**21**].—A cold Me₂CO solution of **18** (60 mg) was treated with Jones reagent, yielding, after vacuum chromatography, 47 mg of **21**: mp 72–75°, [α]²⁵_D 1.5 (c = 0.0135, MeOH); ir ν max cm⁻¹ 1733, 1693, 1240; ¹H nmr (80 MHz, CDCl₃) δ 0.72 (3H, s, Me-18), 0.78 (3H, s, Me-19), 0.91 (3H, d, J = 7 Hz, Me-21), 0.96

(3H, d, $J = 7$ Hz, Me-27), 1.96 (3H, s, OAc), 2.12 (3H, s, OAc), 2.99 (1H, dd, $J = 14, 5$ Hz, H-26ax), 3.75 (1H, d, $J = 14$ Hz, H-26eq), 5.10 (2H, ABq, $J = 23, 13$ Hz, H-Bz), 5.23 (1H, m, H-16 α), 7.30 (5H, bs, aromatic H); ^{13}C nmr see Table 1.

(22*S*,25*R*)-3 β -HYDROXY-16 β -ETHOXY-(22,26)-*N*-Cbz-EPIMINO-5 α -CHOLESTAN-4-ONE [22].— To a solution of 30 mg of **20** in MeOH, a few drops of 5% aqueous K_2CO_3 was added, and the mixture was left overnight at room temperature. Tlc [Si gel plate, C_6H_{14} -EtOAc (1:1)] showed that the 3 β -O-acetate **20**, R_f 0.85, had disappeared and that a new spot with R_f 0.22 was present. The solution was made alkaline and extracted with CHCl_3 . The organic layer was dried over anhydrous Na_2SO_4 and the solvent evaporated to dryness. The residue was crystallized from MeOH yielding 22 mg of **22**: mp 138–142°. Calcd for $\text{C}_{37}\text{H}_{55}\text{NO}_5$, C 74.83, H 9.34, N 2.36; found C 75.12, H 9.47, N 2.28. Ir ν max cm^{-1} 1710, 1694; ^1H nmr (80 MHz, CDCl_3) δ 0.73 (3H, s, Me-18), 0.70 (3H, s, Me-19), 0.87 (3H, d, $J = 6$ Hz, Me-21), 1.00 (3H, d, $J = 6$ Hz, Me-27), 3.01 (1H, dd, $J = 14, 5$ Hz, H-26ax), 3.75 (1H, d, $J = 14$ Hz, H-26eq), 3.95 (1H, m, H-16 α), 4.10 (1H, t, $J = 10$ Hz, H-3 α), 5.10 (2H, ABq, $J = 22, 12$ Hz, H-Bz), 7.3 (5H, bs, aromatic H); ^{13}C nmr see Table 1.

(22*S*,25*R*)-3 β -HYDROXY-16 β -ACETOXY-22,26-*N*-Cbz-5 α -EPIMINO-CHOLESTAN-4-ONE [23].— In a similar manner, 30 mg of **21** was hydrolyzed to yield 23 mg of **23**: mp 151–155°, R_f 0.15 on a Si gel plate [solvent C_6H_{14} -EtOAc (1:1)]. Calcd for $\text{C}_{37}\text{H}_{53}\text{NO}_6$, C 73.11, H 8.79, N 2.30; found C 73.28, H 8.91, N 2.23. Ir ν max cm^{-1} 1735, 1710, 1695, 1245; ^1H nmr (80 MHz, CDCl_3) δ 0.72 (3H, s, Me-18), 0.69 (3H, s, m Me-19), 0.91 (3H, d, $J = 6$ Hz, Me-21), 0.97 (3H, d, $J = 6$ Hz, Me-27), 2.10 (3H, s, OAc), 3.00 (1H, dd, $J = 14, 5$ Hz, H-26ax), 3.75 (1H, d, $J = 14$ Hz, H-26eq), 4.12 (1H, t, $J = 10$ Hz, H-3 α), 5.10 (2H, ABq, $J = 22, 12$ Hz, H-Bz), 5.24 (1H, m, H-16 α), 7.3 (5H, bs, aromatic H); ^{13}C nmr see Table 1.

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